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In chemistry, resonance, also called mesomerism, is a way of describing bonding in certain molecules or ions by the combination of several contributing structures (or forms, also variously known as resonance structures or canonical structures) into a resonance hybrid (or hybrid structure) in valence bond theory. It has particular value for describing delocalized electrons within certain ...

Resonance (chemistry) - Wikipedia

In chemistry, a molecule or ion is called chiral (/ k aɪ ' r æ l /) if it cannot be superposed on its mirror image by any combination of rotations, translations, and some conformational changes. This geometric property is called chirality. The terms are derived from Ancient Greek χεῖρ (cheir), meaning "hand"; which is the canonical example of an object with this property.

Chirality (chemistry) - Wikipedia

Aromatic Rings. Aromatic rings (also known as aromatic compounds or arenes) are hydrocarbons which contain benzene, or some other related ring structure. Benzene, C₆H₆, is often drawn as a ring of six carbon atoms, with alternating double bonds and single bonds. This simple picture has some complications, however. (Everything in organic chemistry has complications!)

Molecule Gallery - Aromatic Rings

Alkanes. Alkanes are hydrocarbons in which the carbon atoms are held together by single bonds. Their general formula is C_nH_{2n+2} for molecules which do not contain ring structures. Alkanes are in some respect the most boring of the organic compounds, since they are unreactive (mostly) towards acids, bases, oxidizing agents, reducing agents, and most of the other reagents that organic ...

Molecule Gallery - Alkanes

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