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controlled process by

which hydrocarbons

are broken down or

rearranged into

smaller, more useful

molecules natural gas

This is typically made

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up of 80% methane,
10% ethane, 4%

propane, and 2%

butane Study Guide:

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Cyclohexane contains
only single bonds and
is representative of the

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simplest type of cyclic hydrocarbons. A ring structure may possess double bonds, as in the following portrayal of the well-known hydrocarbon benzene, which has the composition C_6H_6 . See Figure 8. Figure 8. Benzene.

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Hydrocarbons Answer

Study Guide Key

Concepts 22.1

Hydrocarbons •

Because carbon has four valence electrons, carbon atoms always form four covalent bonds. • The carbon atoms in an alkane can be arranged in a straight chain or in a chain that has branches. • Molecules of hydrocarbons, such as alkanes, are nonpolar molecules.

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22.2 Unsaturated
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Answer

Chemists called the hydrocarbons that reacted with bromine unsaturated

hydrocarbons in the same sense that an unsaturated aqueous solution can dissolve more solute.

Hydrocarbons that did not react with bromine were said to be saturated. Present-day chemists can now

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explain the experimental results obtained 170 years ago.

Chapter 21: Hydrocarbons

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controlled process by

which hydrocarbons

are broken down or

rearranged into

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smaller, more useful
molecules natural gas

This is typically made
up of 80% methane,
10% ethane, 4%
propane, and 2%
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provide a source of

energy and raw

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Chapter 22

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Hydrocarbons are

organic molecules that
only contain two

elements: carbon and
hydrogen. Common

hydrocarbons include

methane, propane, and

butane. Hydrocarbons

are the primary

constituents of fossil...

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